

## Notizen / Notes

Synthesis of  $\beta$ -Oxo Esters from Silyl Enol Ethers and Dichlorobis(phenoxy)methane

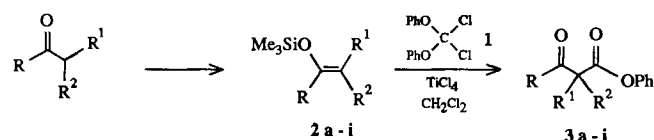
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**Key Words:** Carboxylation / Aryloxyacylation / Siloxyalkenes / Silyl Enol Ethers / Acetals, dichloro /  $\beta$ -Oxo estersDichlorobis(phenoxy)methane (**1**) reacts with trimethylsiloxyalkenes in the presence of 1.2 equiv. of  $\text{TiCl}_4$  to give  $\beta$ -oxo esters in 44–82% yield. The title compound **1** is thus employed as a  $\text{PhO}_2\text{C}^\oplus$  equivalent.

Bis(aryloxy)dichloromethanes (e. g. **1**) are readily available by radical-induced chlorination of formaldehyde diaryl acetals<sup>1)</sup>. Like 2,2-dichloro-1,3-benzodioxol, which is used for the electrophilic carboxylation of aromatic compounds<sup>2)</sup> and of alkenes<sup>3)</sup>, they represent  $\text{RO}_2\text{C}^\oplus$  equivalents and have been employed for the synthesis of  $\beta,\gamma$ -unsaturated esters from allylsilanes<sup>4)</sup>. In this paper we report on the reaction of bis(aryloxy)dichloromethanes with silyl enol ethers under Lewis-acidic conditions yielding  $\beta$ -oxo esters in moderate to good yields. Since silyl enol ethers may regioselectively be synthesized from the corresponding ketones<sup>5)</sup>, the two-step sequence according to eq. (1) represents a useful alternative for the generation of  $\beta$ -oxo esters from ketones, which has previously been carried out under strongly basic conditions using dialkyl carbonates<sup>6)</sup>, dialkyl oxalates<sup>7)</sup>, methyl methoxymagnesium carbonate<sup>8)</sup>, or methyl cyanofornate<sup>9)</sup>.



The experimental conditions of the reaction of **1** with the silyl enol ether **2g** have been optimized. Poor yields of **3g** are obtained, when **1** is treated with **2g** in  $\text{CH}_2\text{Cl}_2$  in the presence of  $\text{ZnCl}_2/\text{Et}_2\text{O}$ <sup>10)</sup> (1.5 equiv.) or in the presence of 0.1 equiv. of  $\text{TiCl}_4$  (<20%). With  $\text{SnCl}_4$  (1.3 equiv.) in  $\text{CH}_2\text{Cl}_2$ , 27% of **3g** is isolated, and an almost quantitative yield of crude **3g** is obtained, when 1.2 equiv. of  $\text{TiCl}_4$  is added to a mixture of **1** and **2g** in  $\text{CH}_2\text{Cl}_2$  at  $-78^\circ\text{C}$ <sup>11)</sup>.

Table 1 (entries a–f) shows that silyl enol ethers, generated from saturated ketones, may efficiently be converted into  $\beta$ -oxo esters according to this method. Only one of the possible diastereoisomers is obtained from compounds **2e** and **2f**. Though some of these compounds are enolizable (**3b–e**), the NMR spectra, taken immediately after dissolving these compounds in  $\text{CDCl}_3$ , indicate the presence of pure keto compounds, and only in the case of **3b**, 10–15% of an enol has been detected. After several days, a ketone-to-enol ratio of 60:40 is observed for **3c**, while the spectra of the other compounds have remained unchanged<sup>12)</sup>.

Table 1. Titanium(IV) chloride promoted reactions of dichlorobis(phenoxy)methane (**1**) with trimethylsiloxyalkenes **2**

Reactants	Products	Yield (%)	IR $\nu(\text{CO})$ [ $\text{cm}^{-1}$ ]
<b>2a</b>	<b>3a</b>	71	1758, 1714
<b>2b</b>	<b>3b</b>	81	1764, 1701
<b>2c</b>	<b>3c</b>	68	1763, 1696
<b>2d</b>	<b>3d</b>	75	1748, 1708
<b>2e</b>	<b>3e</b>	54	1772, 1731
<b>2f</b>	<b>3f</b>	46	1757, 1708
<b>2g</b>	<b>3g</b>	74	1761, 1727
<b>2h</b>	<b>3h</b>	44	1755, 1670
<b>2i</b>	<b>3i</b>	82	1760, 1738

Examples g–i suggest that **1** may also be used for the carboxylation of silyl enol ethers derived from aldehydes (**2g**), of siloxy dienes (**2h**), and ketene silyl acetals (**2i**), but experiments with 1-(trimethylsilyloxy)indene and 1-(trimethylsilyloxy)-1,3-butadiene have not been successful.

## Experimental

IR: IR-435 (Shimadzu). — NMR: XL 200 (Varian), internal standard TMS; for spectra of higher order, virtual couplings are listed. — MS: 70-250E (VG-Instruments).

Dichlorobis(phenoxy)methane (**1**) was prepared by chlorination of bis(phenoxy)methane<sup>11</sup>. The silyl enol ethers **2a–h** were synthesized from the corresponding carbonyl compounds by treatment with Me<sub>3</sub>SiCl, NaI, and (C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>N in CH<sub>3</sub>CN according to the procedures A and D described by Cazeau et al.<sup>13</sup>. The silylated ketene acetal **2i** is commercially available (Fluka).

**Caution:** Because of the known toxicity of chloro ethers, all operations should be carried out in an efficient hood, and skin contact should be avoided.

**Phenyl 2,2-Dimethyl-3-oxobutyrates (3a).** — *General Procedure:* Compounds **1** (0.80 g, 3.0 mmol) and **2a** (0.52 g, 3.3 mmol) were dissolved in 25 ml of CH<sub>2</sub>Cl<sub>2</sub> at –78°C. A solution of TiCl<sub>4</sub> (0.76 g, 4.0 mmol) in 10 ml of CH<sub>2</sub>Cl<sub>2</sub> was added dropwise with stirring (ca. 5 min), and the mixture was kept at –78°C. After 10 h, the dark brown solution was washed with cold 3% aqueous HCl to give a colorless mixture. The organic layer was concentrated to yield an oily residue which was dissolved in pentane and repeatedly washed with 5% aqueous Na<sub>2</sub>CO<sub>3</sub> solution to remove phenol. After drying with Na<sub>2</sub>CO<sub>3</sub> and evaporation of pentane, the residue was purified by layer chromatography [silica gel, hexane/ether (95:5)] yielding 0.51 g of product. Distillation [73–76°C (bath)/0.1 mbar] afforded 0.44 g (71%) of **3a**. — <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 1.52 (s, 6H), 2.30 (s, 3H), 7.05–7.40 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 21.86 (q, 2-CH<sub>3</sub>), 25.77 (q, C-4), 56.02 (s, C-2), 121.11 (d, C<sub>o</sub>), 126.10 (d, C<sub>p</sub>), 129.49 (d, C<sub>m</sub>), 150.50 (s, C<sub>i</sub>), 172.18 (s, C-1), 205.32 (s, C-3). — MS (70 eV): *m/z* (%) = 206 (28) [M<sup>+</sup>], 164 (34), 113 (51), 94 (100), 85 (56), 70 (53), 57 (31), 43 (98).

C<sub>12</sub>H<sub>14</sub>O<sub>3</sub> (206.2) Calcd. C 69.89 H 6.84  
Found C 69.87 H 6.73

**Phenyl 2-Oxocycloheptanecarboxylates (3b):** A mixture of compounds **1** (0.80 g, 3.0 mmol) and **2b** (0.61 g, 3.3 mmol) was treated with 0.76 g (4.0 mmol) of TiCl<sub>4</sub> as described before and worked up after 20 h. Layer chromatography [silica gel, hexane/ether (3:1)] afforded 0.62 g of **3b**, which was distilled to yield 0.56 g (81%) of an analytically pure material with bp 110–115°C (bath)/0.1 mbar. — <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 1.35–2.13, 2.17–2.34, 2.47–2.46 (3 m, 10H), 3.84 (dd, *J* = 10.1 Hz, *J* = 3.7 Hz, 0.9H, keto tautomer), 7.08–7.48 (m, 5H), 12.53 (br. s, 0.1 H, enol tautomer). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 24.04, 27.47, 28.22, 29.45, 43.29 (5 t, C-3, C-4, C-5, C-6, C-7), 58.67 (d, C-1), 121.45 (d, C<sub>o</sub>), 125.94 (d, C<sub>p</sub>), 129.38 (d, C<sub>m</sub>), 150.57 (s, C<sub>i</sub>), 169.35 (s, CO<sub>2</sub>Ph), 208.66 (s, C-2); signals for the enol species (<15%): δ = 24.56, 24.63, 31.95, 35.61 (t), 101.21 (s), 121.79 (d), 125.74 (d), 129.51 (d), 150.49 (s), 172.12 (s). — MS (70 eV): *m/z* (%) = 232 (1) [M<sup>+</sup>], 139 (58), 94 (100).

C<sub>14</sub>H<sub>16</sub>O<sub>3</sub> (232.3) Calcd. C 72.39 H 6.94  
Found C 72.20 H 7.02

**Phenyl 2-Oxocyclooctanecarboxylates (3c):** Compound **1** (1.2 g, 4.5 mmol) was treated with **2c** (1.0 g, 5.1 mmol) in the presence of TiCl<sub>4</sub> (1.1 g, 5.8 mmol) for 18 h to give **3c**. Distillation [125–130°C (bath)/0.2 mbar] afforded 0.75 g (68%) of oily **3c** which solidified;

mp 70–72°C (pentane/ether). — <sup>1</sup>H NMR (CDCl<sub>3</sub>, 40°C): δ = 1.25–2.27, 2.48–2.76 (2 m, 12H), 3.86 (dd, *J* = 8 Hz, *J* = 7 Hz, 1H), 7.01–7.40 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 24.49, 24.66, 24.74, 27.30, 29.48, 42.36 (6 t, C-3, C-4, C-5, C-6, C-7, C-8), 56.34 (d, C-1), 121.34 (d, C<sub>o</sub>), 125.97 (d, C<sub>p</sub>), 129.38 (d, C<sub>m</sub>), 150.45 (s, C<sub>i</sub>), 168.81 (s, CO<sub>2</sub>Ph), 211.71 (s, C-2); signals for the enol species: δ = 24.06, 26.05, 26.54, 28.77, 30.04, 32.59 (6 t, C-3, C-4, C-5, C-6, C-7, C-8), 98.79 (s, C-1), 121.78 (d, C<sub>o</sub>), 125.77 (d, C<sub>p</sub>), 171.70 (s, CO<sub>2</sub>Ph), 178.60 (s, C-2). — MS (70 eV): *m/z* (%) = 246 (1) [M<sup>+</sup>], 153 (100), 94 (32), 55 (54).

C<sub>15</sub>H<sub>18</sub>O<sub>3</sub> (246.3) Calcd. C 73.15 H 7.37  
Found C 73.13 H 7.27

**Phenyl 2-Oxocyclododecanecarboxylates (3d):** Compound **1** (0.80 g, 3.0 mmol) was treated with **2d** (0.84 g, 3.3 mmol) and TiCl<sub>4</sub> (0.76 g, 4.0 mmol) for 24 h according to the described procedure. The residue, obtained after evaporation of pentane (0.90 g), was recrystallized from hexane to give 0.67 g (75%) of **3d** with mp 81–83°C. — <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 1.05–2.08, 2.10–2.56, 2.62–2.91 (3 m, 20H), 3.87 (dd, *J* = 11.0 Hz, *J* = 3.7 Hz, 1H), 6.97–7.46 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 22.01, 22.31, 22.83, 23.85, 24.07 (high intensity), 25.26, 25.58, 26.77, 38.38, (9 t, C-3–C-12), 57.57 (d, C-1), 121.24 (d, C<sub>o</sub>), 126.07 (d, C<sub>p</sub>), 129.46 (d, C<sub>m</sub>), 150.39 (s, C<sub>i</sub>), 168.51 (s, CO<sub>2</sub>Ph), 206.06 (s, C-2). — MS (70 eV): *m/z* (%) = 302 (1) [M<sup>+</sup>], 209 (100), 94 (23), 77 (23), 55 (27), 41 (21).

C<sub>19</sub>H<sub>26</sub>O<sub>3</sub> (302.4) Calcd. C 75.46 H 8.67  
Found C 75.37 H 8.57

**Phenyl Camphor-3-carboxylates (3e)** was synthesized by reaction of **1** (1.2 g, 4.5 mmol) with **2e** (1.1 g, 4.9 mmol) in the presence of TiCl<sub>4</sub> (1.1 g, 5.8 mmol) at –78°C for 20 h. Chromatographic purification of the product [silica gel, hexane/ether (4:1)] gave 0.65 g (54%) of **3e** which was recrystallized from hexane; mp 69–70°C. — <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 0.94, 0.98, 1.06 (3 s, 9H), 1.48–2.07 (m, 4H), 2.60 (br. t, *J* = 4.4 Hz, 1H), 3.58 (dd, *J* = 1.9 Hz, *J* = 4.7 Hz, 1H), 7.00–7.43 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 9.55 (q, 1-CH<sub>3</sub>), 18.79, 19.54 (2 q, 7-CH<sub>3</sub>), 22.52 (t, C-5), 29.40 (t, C-6), 45.88 (d, C-4), 47.23 (s, C-7), 55.50 (d, C-3), 58.59 (s, C-1), 121.46 (d, C<sub>o</sub>), 126.01 (d, C<sub>p</sub>), 129.41 (d, C<sub>m</sub>), 150.33 (s, C<sub>i</sub>), 168.17 (s, CO<sub>2</sub>Ph), 210.81 (s, C-2). — MS (70 eV): *m/z* (%) = 272 (5) [M<sup>+</sup>], 179 (100), 151 (45), 123 (18), 94 (20), 83 (37), 55 (17), 41 (24).

C<sub>17</sub>H<sub>20</sub>O<sub>3</sub> (272.3) Calcd. C 74.97 H 7.40  
Found C 74.95 H 7.46

**Phenyl 1-Methyl-4-(1-methylvinyl)-2-oxocyclohexanecarboxylates (3f)** was prepared by treatment of **1** (0.80 g, 3.0 mmol) with **2f** (0.74 g, 3.3 mmol), and TiCl<sub>4</sub> (0.76 g, 4.0 mmol) at –78°C for 17 h. Purification of the product by column chromatography [silica gel, hexane/ether (95:5)] gave 0.45 g of **3f** which was further purified by distillation yielding 0.37 g (46%) of product with bp 130 to 135°C/0.15 mbar. — <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ = 1.53 (s, CH<sub>3</sub>), 1.74 to 2.05 (m, superimposed by br. s at 1.77, 7H), 2.42–2.58 (m, 1H), 2.63 (m, 2H), 4.76, 4.86 (2 br. s, 2H), 7.05–7.45 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ = 20.48, 21.20 (2 q, 1-CH<sub>3</sub>, CH<sub>3</sub>C=), 25.40 (t, C-5), 33.68 (t, C-6), 43.35 (t, C-3), 44.22 (d, C-4), 57.19 (s, C-1), 111.27 (t, CH<sub>2</sub>=), 121.40 (d, C<sub>o</sub>), 125.96 (d, C<sub>p</sub>), 129.42 (d, C<sub>m</sub>), 145.48 (s, C=CH<sub>2</sub>), 150.73 (s, C<sub>i</sub>), 171.92 (s, CO<sub>2</sub>Ph), 208.34 (s, C-2). — MS (70 eV): *m/z* (%) = 272 (12) [M<sup>+</sup>], 179 (83), 151 (15), 123 (30), 109 (51), 107 (55), 97 (100), 94 (41), 81 (51), 69 (38), 67 (30), 55 (72), 41 (60).

C<sub>17</sub>H<sub>20</sub>O<sub>3</sub> (272.3) Calcd. C 74.97 H 7.40  
Found C 74.92 H 7.37

**Phenyl 2,2-Dimethyl-3-oxopropanoates (3g):** The crude material obtained from **1** (1.2 g, 4.5 mmol), **2g** (0.79 g, 5.5 mmol), and TiCl<sub>4</sub> (1.1 g, 5.8 mmol) according to the general procedure was purified

by column chromatography [silica gel, hexane/ether (4:1)] to yield 0.75 g of **3g**. Distillation [84–88°C (bath)/0.4 mbar] afforded 0.63 g (74%) of analytically pure **3g**. — <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  = 1.50 (s, 6H), 7.05–7.41 (m, 5H), 9.80 (s, 1H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  = 19.71 (q, 2-CH<sub>3</sub>), 54.03 (s, C-2), 121.24 (d, C<sub>o</sub>), 126.21 (d, C<sub>p</sub>), 129.51 (d, C<sub>m</sub>), 150.32 (s, C<sub>i</sub>), 171.30 (s, C-1), 198.50 (d, C-3). — MS (70 eV):  $m/z$  (%) = 192 (12) [M<sup>+</sup>], 164 (25), 99 (49), 94 (100), 71 (47), 70 (43), 43 (82).

C<sub>11</sub>H<sub>12</sub>O<sub>3</sub> (192.2) Calcd. C 68.74 H 6.29  
Found C 68.70 H 6.33

*Phenyl (5,5-Dimethyl-3-oxo-1-cyclohexen-1-ylacetate) (3h)*: Compound **1** (0.80 g, 3.0 mmol) was treated with **2h** (0.70 g, 3.3 mmol) and TiCl<sub>4</sub> (0.76 g, 4.0 mmol) for 18 h to give crude **3h** which was purified by column chromatography [silica gel, ether/hexane (2:1)] and distillation [145–150°C (bath)/0.08 mbar] yielding 0.40 g of product. Recrystallization from hexane afforded 0.34 g (44%) of **3h** with mp 68–69°C. — <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  = 1.09 (s, 6H), 2.28 (s, 2H), 2.38 (br. s, 2H), 3.47 (br. s, 2H), 6.09 (br. s, 1H), 7.05–7.50 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  = 28.13 (q, 5-CH<sub>3</sub>), 33.73 (s, C-5), 43.12, 43.56 (2 t, C-6, O<sub>2</sub>CCH<sub>2</sub>), 50.85 (t, C-4), 126.26 (d, C<sub>o</sub>), 126.13 (d, C<sub>p</sub>), 128.17 (d, C-2), 129.49 (d, C<sub>m</sub>), 150.29 (s, C<sub>i</sub>), 154.01 (s, C-1), 167.86 (s, CO<sub>2</sub>Ph), 199.39 (s, C-3). — MS (70 eV):  $m/z$  (%) = 258 (4) [M<sup>+</sup>], 165 (100), 109 (12), 108 (25), 94 (22), 77 (7), 67 (10), 53 (10).

C<sub>16</sub>H<sub>18</sub>O<sub>3</sub> (258.3) Calcd. C 74.40 H 7.02  
Found C 74.31 H 7.10

*Methyl Phenyl (2,2-Dimethylmalonate) (3i)*: The product generated from **1** (0.80 g, 3.0 mmol) and **2i** (0.58 g, 3.3 mmol) in the presence of TiCl<sub>4</sub> (0.76 g, 4.0 mmol) within 14 h was purified by layer chromatography [silica gel, hexane/ether (95:5)] to yield 0.60 g of **3i**. Distillation [74–78°C (bath)/0.1 mbar] afforded 0.54 g (82%) of analytically pure **3i**. — <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  = 1.59 (s, 6H), 3.81 (s, 3H), 7.04–7.45 (m, 5H). — <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  = 22.75 (q, 2-CH<sub>3</sub>), 50.08 (s, C-2), 52.69 (q, OCH<sub>3</sub>), 121.21 (d, C<sub>o</sub>), 125.99 (d, C<sub>p</sub>), 129.43 (d, C<sub>i</sub>), 150.67 (s, C<sub>i</sub>), 171.36, 172.99 (2 s, CO<sub>2</sub>CH<sub>3</sub>, CO<sub>2</sub>Ph). — MS (70 eV):  $m/z$  (%) = 222 (30) [M<sup>+</sup>], 129 (97), 101 (100), 94 (80), 73 (45), 41 (27).

C<sub>12</sub>H<sub>14</sub>O<sub>4</sub> (222.2) Calcd. C 64.85 H 6.35  
Found C 64.48 H 6.57

## CAS Registry Numbers

**1**: 4885-03-4 / **2a**: 17510-44-0 / **2b**: 22081-48-7 / **2c**: 50338-42-6 / **2d**: 51584-36-2 / **2e**: 56613-17-3 / **2f**: 72311-10-5 / **2g**: 6651-34-9 / **2h**: 80239-27-6 / **2i**: 31469-15-5 / **3a**: 103439-34-5 / **3b**: 126256-23-3 / **3c**: 126256-24-4 / **3d**: 126256-25-5 / **3e**: 126256-26-6 / **3f**: 126256-27-7 / **3g**: 89635-72-3 / **3h**: 126256-28-8 / **3i**: 126256-29-9

- <sup>1)</sup> A. Cambanis, E. Bäuml, H. Mayr, *Synthesis* **1988**, 961; Bayer AG (H. Mayr, A. Cambanis, E. Bäuml, Inv.), D.A.S. 3821 130.0.
- <sup>2)</sup> H. Gross, J. Rusche, M. Mirsch, *Chem. Ber.* **96** (1963) 1382.
- <sup>3)</sup> H. Mayr, U. von der Brüggen, *Chem. Ber.* **121** (1988) 339.
- <sup>4)</sup> H. Mayr, A. Cambanis, E. Bäuml, *Synthesis*, **1988**, 962.
- <sup>5)</sup> <sup>5a)</sup> E. W. Colvin, *Silicon Reagents in Organic Synthesis*, p. 99, Academic Press, London 1988. — <sup>5b)</sup> E. W. Colvin, *Silicon in Organic Synthesis*, p. 198, Butterworths, London 1981. — <sup>5c)</sup> W. P. Weber, *Silicon Reagents for Organic Synthesis*, p. 255, Springer Verlag, Berlin 1983.
- <sup>6)</sup> <sup>6a)</sup> A. P. Krapcho, J. Diamanti, C. Cayen, R. Bingham, *Org. Synth., Coll. Vol. V* (1973) 198. — <sup>6b)</sup> S. B. Soloway, F. B. La-Forge, *J. Am. Chem. Soc.* **69** (1947) 2677.
- <sup>7)</sup> C. R. Hauser, F. W. Swamer, J. T. Adams, *Org. React.* **8** (1954) 59.
- <sup>8)</sup> <sup>8a)</sup> M. Stiles, *J. Am. Chem. Soc.* **81** (1959) 2598. — <sup>8b)</sup> S. W. Pelletier, R. L. Chappell, P. C. Parthasarthy, N. Lewin, *J. Org. Chem.* **31** (1966) 1747. — <sup>8c)</sup> A. Pavia, F. Winternitz, R. Wylde, *C. R. Acad. Sci.* **261** (1965) 1026. — <sup>8d)</sup> S. Julia, C. Huynh, *C. R. Acad. Sci., Ser. C*, **270** (1970) 1517.
- <sup>9)</sup> L. N. Mander, S. P. Sethi, *Tetrahedron Lett.* **24** (1983) 5425.
- <sup>10)</sup> H. Mayr, W. Striepe, *J. Org. Chem.* **50** (1985) 2995.
- <sup>11)</sup> Similar conditions have previously been applied for the tertiary alkylation of silyl enol ethers: M. T. Reetz, W. F. Maier, H. Heimbach, A. Giannis, G. Anastassiou, *Chem. Ber.* **113** (1980) 3734.
- <sup>12)</sup> Detailed studies on keto-enol equilibria of the corresponding ethyl and methyl esters have been described: <sup>12a)</sup> S. J. Rhoads, C. Pryde *J. Org. Chem.* **30** (1965) 3212. — <sup>12b)</sup> S. J. Rhoads *J. Org. Chem.* **31** (1966) 171. — <sup>12c)</sup> K. R. Kallury, U. J. Krull, M. Thompson *J. Org. Chem.* **53** (1988) 1320.
- <sup>13)</sup> P. Cazeau, F. Duboudin, F. Moulines, O. Babot, J. Dunogues, *Tetrahedron* **43** (1987) 2075.

[45/90]